Before discussing the rejection in detail, a brief review of the presently claimed

invention may be quite instructive. The subject invention relates to a film for metallization

which is composed of a polypropylene-based resin composition for metallized films. The

composition comprises, among other things, a propylene random copolymer (A) produced

in the presence of a metallocene catalyst, and which has the properties (a-1) to (a-6) as

recited in claim 1. In addition, the film for metallization composed of a polypropylene-

based resin composition of independent claim 1 satisfies a mathematical relationship

between the heat seal temperature and the tensile modulus of the film as set forth in

formula (a).

Such a film in accordance with the above provides a metallized film excellent in

processability, stiffness, heat-sealing property, resistance to blocking and surface

scratching, containing a limited quantity of solubles, and excellent in adhesion properties

to the metallizing film and printability and delamination characteristics of the metallized

surface. Also, with the improved stiffness and heat sealing property, the metallized films

in accordance with the presently claimed invention are particularly adapted for use in

wrapping food and medical products. Therefore, it is quite important that copolymer (A)

satisfies all of requirements (a-1) to (a-6) and formula (a), since, even if just one of these

requirements is not satisfied, the resultant film is not suitable as a surface to be metallized.

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It is submitted that the subject films for metallization composed of a polypropylene-

based resin composition as recited in independent claim 1 patentably distinguish over the

films taught by the cited Chatteriee patent. Specifically, as was asserted in the prior

Response, the subject polypropylene-based resin compositions as claimed distinguish over

that taught by the cited Chatterjee patent by the recitation that the film satisfies the

mathematical relationship according to formula (a) between the heat seal temperature at

a specific load and tensile modulus. Further, it is urged, as in the previous Response, that

it is extremely important that copolymer (A) satisfies all of requirements (a-1) to (a-6)

because, if even one of requirements (a-1) to (a-6) is not satisfied, the resultant

composition is not suitable as a material of a film for metallizing thereof.

It was asserted in the subject Action that the above arguments was that they were

not persuasive. More particularly, it now has been asserted that (1) the compositions of

resin according to the cited Chatterjee patent are substantially identical to those as claimed

and (2) both the polypropylene polymer of the Chatterjee patent and the polypropylene

polymer as claimed can be prepared by polymerization methods using substantially

identical magnesium chloride supported titanium-based catalysts. Consequently, it was

concluded that the heat-sealing properties as claimed are inherently possessed by the

compositions of the patent.

As to position (2) above, it is submitted that this position is in error as, among other

things, a particular recitation contained in claim 1 has been conveniently ignored.

Specifically, recitation (A) of claim 1 which recites a propylene random copolymer produced

in the presence of a metallocene catalyst has not been considered. On the other hand,

the only catalyst mentioned in the Chatterjee patent is a titanium-containing polymerization

catalyst system.

More specifically, it was stated in the outstanding Office Action that the propylene

polymer of the Chatterjee patent can be produced using a magnesium chloride supported

titanium-based catalyst which is substantially the same as that used in the presently

claimed invention. However, it is submitted that this assertion is not supported by the

teachings of the subject application.

It is acknowledged that, as set forth in the Action, Production Example 5 of the

subject application relates to a propylene polymer (PP5) which is produced using a

magnesium chloride supported titanium-based catalyst. However, in Production Example

5, the titanium compound supported on magnesium chloride is titanium tetra-n-butoxide

(Ti(O-n-C₄H₉)₄), which is not a metallocene compound. Further, PP5 is used in

Comparative Example 10 (see Table 3) which describes results of a control experiment,

and the properties of the film produced using PP5 are poor (See Table 4). Therefore, PP5

is not a polypropylene-based resin composition film in accordance with the presently

claimed invention.

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As apparent from the above, the magnesium chloride supported titanium-based

catalyst mentioned in the subject application cannot be used for producing the propylene

polymer as a material of the film of the presently claimed invention. Therefore, the film of

the presently claimed invention cannot be rendered obvious from the Chatterjee patent.

In addition, it also was stated in the Action that the catalyst taught by the Chatterjee

patent generically include a Ziegler-Natta type catalyst and metallocene catalyst. However,

as set forth above, the Chatterjee patent only mentions a titanium-containing

polymerization catalyst system. There is no description in the **Chatterjee** patent disclosing

or suggesting a metallocene catalyst. Consequently, a person of ordinary skill in the art

would be led to believe that a metallocene catalyst is excluded from the catalyst taught by

the **Chatterjee** patent.

For the sake of argument, even if the Chatterjee patent is regarded as suggesting

a metallocene catalyst, it cannot be assumed that the propylene polymers produced

thereby would satisfy all requirements of the presently claimed invention based only on the

suggested use of metallocene catalyst. In support thereof, it is to be emphasized that,

according to the presently claimed invention, not all propylene random copolymers

produced in the presence of a metallocene catalyst satisfy all of requirements (a-1) to (a-6).

For example, although the catalyst used in producing Production Example 8 is a

metallocene compound, the resultant propylene polymer (PP8) is used in Comparative

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Example 13. In particular, Table 3 describes results of a control experiment, and the

properties of the film produced using PP8 are poor as is set forth in Table 4. Thus, the

present application clearly demonstrates that requirements (a-1) to (a-6) do not necessarily

follow from the production (polymerization) of a propylene polymer with a metallocene

catalyst.

Therefore, even if it is assumed that the <u>Chatterjee</u> patent suggests a metallocene

catalyst, an assumption not well founded, the use of the metallocene catalyst does not

inherently provide a propylene polymer simultaneously satisfying requirements (a-1) to (a-

6). As a consequence, a person of ordinary skill in the art would not be led to a propylene

polymer useful for producing a metallized film which is produced using a metallocene

catalyst and simultaneously satisfies requirements (a-1) to (a-6).

Second, the page from the Declaration accompanying the previously submitted

Amendment provided a value of 17[HST] - [YM] for three films made of resin compositions

according to the Chatterjee patent (Examples A, D and E) and for a film in accordance with

the present invention (Example 1). The values for each film were set forth in the

handwritten table at the lower right of the page from the Declaration submitted with the last

Amendment.

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Again, it is to be noted that amended claim 1 recites that the value of 17[HST] - [YM]

should be between 1165 and 1670. The page of the Declaration enclosed with the

Amendment indicated that the film of Example A of the Chatterjee patent had a value of

1790; that the film of Example B of the patent had a value of 1752; and that the film of

Example C of the patent had a value of 1698. In distinct contrast, the film in accordance

with the presently claimed invention and as set forth in Example 1 had a value of 1453.

As is readily apparent, the three films according to the patent do not satisfy formula (a), but

the film of Example 1 does satisfy the formula. Thus, it is submitted that the information

from the Declaration disproves the assertion of the examiner that the heat-sealing

properties as claimed are inherently possessed by the compositions of the Chatterjee

patent.

For the reasons stated above, withdrawal of the rejection under 35 U.S.C. § 102(b)

or 35 U.S.C. § 103(a) and allowance of claims 1, 3 and 7 over the cited Chatteriee patent

are respectfully requested.

In view of the foregoing, it is submitted that the subject application is now in

condition for allowance and early notice to that effect is earnestly solicited.

In the event this paper is not timely filed, the undersigned hereby petitions for an

appropriate extension of time. The fee for this extension may be charged to Deposit .

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Account No. 01-2340, along with any other additional fees which may be required with respect to this paper.

Respectfully submitted,

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